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ChemQuest 8 PBworks

October 27th, 2018 - Information Atoms Ions Masses of Subatomic Particles
The atomic mass unit amu is a special unit for measuring the mass of very small particles such as atoms The relationship between amu and grams is the following $1.00 \text{ amu} = 1.66 \times 10^{-24} \text{ g}$ Note the following diagrams of a fluorine atom and a fluorine ion

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